# Hesperetin: A Potent Antioxidant Against Peroxynitrite

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Peroxynitrite  $(ONOO^-)$  is a reactive oxidant formed from superoxide  $(\overline{O_2})$  and nitric oxide (NO), that can oxidize several cellular components, including essential protein, non-protein thiols, DNA, low-density lipoproteins (LDL),<br>and membrane phospholipids. ONOO<sup>-</sup> has contributed to the pathogenesis of diseases such as stroke, heart disease, Alzheimer's disease, and atherosclerosis. Because of the lack of endogenous enzymes to thwart ONOO<sup>-</sup> activation, developing a specific  $ONOO^-$  scavenger is remarkably important. In this study, the ability of hesperetin  $(3^{\prime},5,\overline{7})$ trihydroxy-4-methoxyflavanone) to scavenge  $ONOO^-$  and to protect cells against ONOO<sup>-</sup> and ROS was investigated. The data gained show that hesperetin can efficiently scavenge authentic ONOO<sup>-</sup>. In spectrophotometric analysis, the data revealed that hesperetin led to declined ONOO<sup>-</sup>-mediated nitration of tyrosine through electron donation. Hesperetin exhibited significant inhibition on the nitration of bovine serum albumin (BSA) by  $ONOO^-$  in a dose-dependent manner. Hesperetin also manifested cytoprotection from cell damage induced by  $ONOO<sup>-</sup>$  and ROS. The present study suggests that hesperetin is a powerful  $\rm ONOO^-$  scavenger and promotes cellular defense activity in the protection against ONOO<sup>-</sup> involved diseases.

Keywords: Hesperetin (3',5,7-trihydroxy-4-methoxyflavanone); Nitration; Nitrotyrosine; Peroxynitrite; Cytoprotection

# INTRODUCTION

Flavonoids are a large group of dietary polyphenolic compounds that generally consist of flavanols, flavonols, flavonones, isoflavones, and catechins. Among these compounds, flavonones are widely distributed in the citrus species. Hesperidin, one of the most abundant flavonoids found in citrus fruits,

such as lemons, oranges, and grapefruits, [1,2] has been exhibited to exert pharmacological effects, including capillary permeability<sup>[3]</sup> and anti-carcinogenic, anti-tumor, anti-inflammatory, and anti-oxidant effects. $[4-7]$  Hesperetin is derived from the hydrolysis of aglycone, hesperidin (hesperetin 7 rhammnoglucoside).[8]

Reactive oxygen and nitrogen species (ROS/RNS) generated from various cells, including macrophages and neutrophils, have been reported to be effective materials against various infectious pathogens.<sup>[9]</sup> In turn, superoxide  $(O_2^-)$  may dismutate, either spontaneously or catalyzed by superoxide dismutases (SOD), to form hydrogen peroxide, which then possibly generates other  $\text{ROS}^{\left[10\right]}$  Nitric oxide (NO) is a unique diffusible molecular messenger in the vascular and nervous systems, and is produced by a family of enzymes called nitric oxide synthase (NOS) through the enzymatic oxidation of the guanidine group of L-arginine.[11]

Concurrent generation of NO and  $O_2^-$  can produce a cytotoxic reaction product, peroxynitrite  $\text{CONOO}^{-}$ ).<sup>[12-14]</sup> ONOO<sup>-</sup> is a strong oxidant and nitrating species, the formation of which may be beneficial in inflammatory reactions in terms of oxidative destruction of intruding microorganisms.<sup>[15]</sup> Steady-state concentrations of ONOO<sup>-</sup> may be significant in the vicinity of activated macrophages that concomitantly generate  $\cdot O_2^-$  and  $2\text{NO}$ .<sup>[10]</sup> Despite its non-radical nature, ONOO<sup>-</sup> is more reactive than its parent molecules. ONOO<sup>-</sup> initiates lipid peroxidation, causes DNA breakage, and reacts with thiols.<sup>[16]</sup> ONOO<sup>-</sup> induced modification of proteins has been connected with

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the development of several diseases, including Alzheimer's disease, Huntington's Chorea, and Parkinson's disease.<sup>[17]</sup> ONOO<sup>-</sup> induced protein modifications include protein oxidation (on methionine, cysteine, tryptophane, or tyrosine residues) and nitration (of tyrosine or tryptophane residues).<sup>[16]</sup> The nitration of tyrosyl residues in proteins may interfere with signaling pathways relying on tyrosine phosphorylation/dephosphorylation, which regulate cellular proliferation, apoptosis,<sup>[18]</sup> and oxidation. Also, the nitration of proteins mediated by  $ONOO^-$  may disrupt the tertiary structure of proteins and, at higher degrees of damage, may render them inaccessible for degradation by the proteasome.<sup>[19]</sup> Moreover, ONOO<sup>-</sup> ability to modify low-density lipoproteins (LDL) may initiate cellular signals that can lead to inflammation, mitosis or cholesterol accumulation, thereby resulting in atherosclerosis.<sup>[20]</sup> Because endogenous scavenging enzymes that can inactive  $ONOO^-$  are lacking, investigating specific scavengers to  $ONOO^-$  is considerably important. The selenium-containing compounds,  $D(-)$ penicillamine<sup>[21]</sup> and ebselen,<sup>[22]</sup> have been shown to be a powerful scavenger against ONOO<sup>-</sup>. Recently, there is an increasing interest in screening natural products for possible  $ONOO^-$  scavengers.

The goal of this study is to define the effectiveness of hesperetin as an antioxidant. In doing this, the possible regulation of  $ONOO^-$  and its major production pathway, by hesperetin was checked using several *in vitro* models and *tert*-butylhydroperoxide (t-BHP)-stimulated YPEN-1 cells. Our results suggested that hesperetin may be a selective regulator of ONOO<sup>-</sup>-mediated diseases via its direct scavenging activity.

# MATERIALS AND METHODS

# **Materials**

Hesperetin (3',5,7-trihydroxy-4-methoxyflavanone) (Fig. 1), DL-penicillamine (DL-2-amino-3-mercapto-3 methylbutanoic acid), bovine serum albumin (BSA), and t-BHP were obtained from Sigma Chemical Co. (St Louis, MO, USA). Trolox (6-hydroxy-2-carboxylic acid) was obtained from Aldrich Chemical Co. (Milwaukee, WI, USA). Dihydrorhodamine 123 (DHR 123) and 2,7-di-chlorodihydrofluorescein diacetate  $(H<sub>2</sub>DCFDA)$  were from Molecular Probes (Eugene, OR, USA) and  $ONOO^-$  was from Calbiochem Co. (San Diego, CA, USA). 4,5-Di-aminofluorescein (DAF-2) was purchased from Daiichi Pure Chemicals Co. (Tokyo, Japan). Polyvinylidene fluoride (PVDF) membrane (Immobilon-P) was obtained from Millipore Corp. (Billelica, MA, USA), and the chemiluminescence detection system was



FIGURE 1 Chemical structure of hesperetin.

from Amersham Life Sciences, Inc. (Arlington Heights, IL, USA). Antibody to nitrotyrosine was from Upstate Biotechnology (Lake Placid, NY, USA). Enhanced chemiluminescence (ECL) Western blotting detection reagents were from Amersham Life Science (Buckinghamshire, UK). All other chemicals were of the highest purity available from either Sigma Chemical Co. (St Louis, MO, USA) or Junsei Chemical Co. (Tokyo, Japan).

#### Measurement of ONOO<sup>-</sup> Scavenging Activity

 $ONOO^-$  scavenging was measured by monitoring the oxidation of DHR 123 by modifying the method of Kooy et al.<sup>[23]</sup> A stock solution of 5 mM DHR 123 in dimetyhlformamide was purged with nitrogen and stored at  $-20^{\circ}$ C. Working solution with 5  $\mu$ M DHR 123 diluted from the stock solution was placed on ice in the dark immediately before the experiment. The buffer (90 mM sodium chloride, 50 mM sodium phosphate (pH 7.4), and 5 mM potassium chloride) was purged with nitrogen and placed on ice before use. Just before use, 5 mM diethylenetriaminepentaacetic acid (DTPA) was added. The ONOO<sup>-</sup> scavenging ability, by the oxidation of DHR 123, was measured at room temperature on a microplate fluorescence Genious (Tecan, Austria) with excitation and emission wavelengths of 485 and 535 nm, respectively. The background and final fluorescent intensities were determined 5 min after treatment with or without authentic  $10 \mu M ONOO^-$  in 0.3 N sodium hydroxide (NaOH). Authentic ONOO<sup>-</sup> easily oxidized DHR 123 with its final fluorescent intensity being stable over time. Penicillamine was used as a positive control.

# Measurement of  $\cdot O_2^-$  Scavenging Activity

H2DCFDA (f.c. 2.5 mM) mixed with esterase (f.c. 1.5 units/ml) was incubated at  $37^{\circ}$ C for 20 min and placed on ice in the dark until prior to the study. H2DCFDA was oxidized to fluorescent 2,7-dichlorofluorescein (DCF) by  $\cdot O_2^-$ . A phosphate buffer (50 mM) at pH 7.4 was used. The fluorescence intensity of DCF was measured for 1h by using microplate fluorescence with excitation and emission wavelengths of 485 and 535 nm, respectively, with or without the addition of menadione (50 mM) as an  $\cdot$ O<sub>2</sub> source.

# Measurement of ·NO Scavenging Activity

·NO scavenging activity was determined by monitoring DAF-2 by modifying the method of Chung *et al.*<sup>[24]</sup> DAF-2, as a specific  $\cdot$ NO indicator, selectively traps NO between two amino groups in its molecule, and yields triazolofluorescein, which emits green fluorescence when excited at 490-495 nm.<sup>[25]</sup> In brief, 1 mg DAF-2 in 0.55 ml of dimethyl sulfoxide was diluted with 50 mM phosphate buffer (pH 7.4) to 1/400 fold.

A  $\cdot$ NO donor, 2 mM sodium nitroprusside (SNP), and  $3.14 \mu M$  DAF-2 were added to a 96-well black microplate. The fluorescence intensity was dependent on the amount of  $\cdot$ NO trapped by DAF-2. The fluorescence signal caused by the reaction of DAF-2 with  $\cdot$ NO was measured for 30 min by the fluorescence at excitation and emission wavelengths of 495 and 515 nm, respectively.

# Inhibition of ONOO<sup>-</sup>-mediated Tyrosine Nitration by Hesperetin

The ability of hesperetin to inhibit the formation of 3-nitrotyrosine was quantified as the index of hesperetin's inhibition of tyrosine nitration utilizing spectrophotometric and Western blot analyses.

# Spectrophotometric Analysis of Nitrated Proteins

The spectral change in the visible region of hesperetin in the presence of  $ONOO^-$  was monitored at 430 nm to detect the existence of nitration.  $ONOO^{-}$  (200  $\mu$ M) in 0.3 N NaOH was added to a solution containing a sample and tyrosine  $(200 \mu M)$ in 50 mM phosphate buffer (pH 7.4), making a final volume of 1 ml. Each mixed solution was scanned between 190 and 600 nm on an Ultraspec 2000 UVvisible spectrophotometer (Pharmacia-Biotech, UK).

# Western Blotting Analysis

A 2.5  $\mu$ l aliquot of hesperetin dissolved in 10% v/v EtOH was added to  $95 \mu l$  of BSA (0.5 mg of protein/ml) with  $2.5 \mu l$  of ONOO<sup>-</sup> (100  $\mu$ M). The mixed samples were incubated with shaking at  $37^{\circ}$ C for 20 min.

The prepared samples in gel buffer, pH 6.8 (12.5 mM tris[hydroxymethyl]aminomethane, 4% sodium dodecyl sulfate (SDS), 20% glycerol, 10% 2-mercaptoethanol and 0.2% bromophenol blue) in a ratio of 1:1 were boiled for 5 min. Total protein equivalents for each sample were separated on 8%

SDS-polyacrylamide minigel at 100 Vand transferred to a PVDF membrane at 100 V for 90 min in a wet transfer system (Bio-Rad, Hercules, CA). The membrane was immediately placed into a blocking solution (5% w/v skim milk powder in TBS-Tween buffer containing 10 mM Tris, 100 mM NaCl, and 0.1 mM Tween-20, pH 7.4) at room temperature for 1 h. The membrane was washed in TBS-Tween buffer for 30 min and then incubated with a monoclonal anti-nitrotyrosine antibody (1% w/v skim milk, diluted 1:2000 in TBS-Tween buffer) at room temperature for 3 h. After three 10 min washing in TBS-Tween buffer, the membrane was incubated with horseradish peroxidase-conjugated anti-mouse secondary antibody from sheep  $(1\% \text{ w/v})$  skim milk, diluted 1:2000 in TBS-Tween buffer) at room temperature for 2 h. After three 10 min washing in TBS-Tween buffer, antibody labeling was detected using ECL and exposed to radiographic film. Prestained blue protein markers were used for molecular-weight determination.

# Cells and Cell Culture Conditions

YPEN-1 cells, rat prostatic endothelial cells, were obtained from ATCC (American Type Culture Collection, Rockville, MD, USA). YPEN cells were cultured in Dulbecco's Modified Eagle Media (DMEM) (Nissui Co., Tokyo, Japan) supplemented with 5% heat-inactivated (56 $\degree$ C for 30 min) fetal bovine serum (Gibco, Grand Island, NY),  $233.6 \,\text{mg/ml}$  glutamine,  $72 \,\mu\text{g/ml}$  penicillin– streptomycin, and  $0.25 \,\mu\text{g/ml}$  amphotericin B, and adjusted to pH 7.4.–7.6 with NaHCO<sub>3</sub> in an atmosphere of  $5\%$  CO<sub>2</sub>. The fresh medium was replaced after one day to remove non-adherent cells or cell debris.

# Determination of Intracellular Free Radical Scavenging Activity

#### Inhibition of Intracellular ROS

YPEN-1 cells in a 96-well plate were pre-incubated for 24 h. After one day, the medium was changed to fresh serum free medium. The cells were treated with or without hesperetin and incubated for 2 h. After treatment with  $t$ -BHP (20  $\mu$ M) for 1 h, the medium was replaced with fresh serum free medium and  $H_2$ DCFDA (125  $\mu$ M) was added. The fluorescence intensity of DCF was measured for 1 h using the microplate fluorescence with excitation and emission wavelengths of 485 and 535 nm, respectively.

For intracellular ROS generation on endothelial cells, various oxygen species, including  $O_2^-$ , hydrogen peroxide, hydroxyl radicals and lipid hydroxides, were observed via a fluorescence microscope at  $\times$  120 magnification using the probe, DCFDA. Cell seeded in a 6-well plate adhered overnight, and then the culture medium was replaced with fresh serum free medium. Cells were exposed to hesperetin for 2 h and then treated with t-BHP (20  $\mu$ M) for 1 h. After treatment, the medium was removed, and cells were added DCFDA (10  $\mu$ M). Then cells were observed via a fluorescence microscope Axiovert 100 (Zeiss, Germany).

# Inhibition of Intracellular ONOO<sup>-</sup>

YPEN-1 cells in a 96-well plate were pre-incubated for 24 h. After pre-incubation with or without hesperetin for 2h, cells were treated with  $20 \mu M$ t-BHP for 1 h. And then the cells were added to the rhodamine solution (50 mM sodium phosphate buffer, 90 mM sodium chloride, 5 mM potassium chloride,  $5 \text{ mM}$  DTPA, and DHR 123). The  $\text{ONOO}^{-}$ scavenging ability by the oxidation of DHR 123 was determined on the microplate fluorescence for 1 h with excitation and emission wavelengths of 485 and 535 nm, respectively.

Intracellular ONOO<sup>-</sup> generation on endothelial cells was observed via a fluorescence microscope at £ 120 magnification using the probe, DHR 123. Cells seeded in a 6-well plate were incubated overnight, and the culture medium was changed to a fresh, serumfree medium. Cells were exposed to hesperetin for 2 h and then treated with *t*-BHP (20 $\mu$ M) for 1 h. After treatment, medium was removed and cells were added to DHR 123 (10  $\mu$ M). Then cells were observed via a fluorescence microscope.

# Statistical Analysis

Data are expressed as mean  $\pm$  standard deviation (SD) of three or five experiments. Statistical analysis was confirmed by Student's *t*-test. Values of  $p < 0.05$ were considered to be statistically significant.

# RESULTS

#### The ONOO<sup>-</sup> Scavenging Activity of Hesperetin

The ability of hesperetin to scavenge  $ONOO<sup>-</sup>$  was investigated using DHR 123 as shown in Table I. The results showed that hesperetin has similar

TABLE I Effect of  $IC_{50}$  ( $\mu$ M) hesperetin on ONOO<sup>-</sup>,  $\cdot$ O<sub>2</sub><sup>-</sup>, and ·NO scavenging activities

	$ONOO^-$	$\cdot O_{2}$	·NO
Hesperetin	$4.67 \pm 0.16$	$7.47 \pm 1.58$	$18.57 \pm 2.47$
Penicillamine* Trolox*	$4.04 \pm 0.03$	$9.80 \pm 0.51$	
Carboxy-PTIO*			$5.38 \pm 0.85$

\* Used as a positive control. IC50: 50% inhibition concentration. Abbreviations: ONOO<sup>-</sup>, peroxynitrite;  $\overline{O_2}$ , superoxide; NO, nitric oxide.

scavenging activity to penicillamine, a well-known, strong ONOO<sup>-</sup> scavenger.

# Scavenging Ability of Hesperetin Against  $\cdot$ O<sub>2</sub> and  $\cdot$ NO

Table I shows IC<sub>50</sub> values of hesperetin for  $\cdot$ O<sub>2</sub><sup>-</sup> and  $\cdot$ NO scavenging abilities. Because ONOO $^-$  can be formed from reaction of  $\cdot$ NO with  $\cdot$ O<sub>2</sub><sup> $-$ </sup>, we thought it is interesting to compare the active components of  $ONOO^-$ ,  $\cdot O_2^-$  and  $\cdot$ NO. Compared with trolox (IC<sub>50</sub>:  $9.80 \pm 0.51$  mM), a well-known antioxidant  $\cdot$ O<sub>2</sub> <sup>2</sup> scavenger, hesperetin showed higher scavenging activity levels (IC<sub>50</sub>:  $7.47 \pm 1.58 \text{ mM}$ ). In addition, hesperetin inhibitory action was slight (IC<sub>50</sub>:  $18.57 \pm 2.47$  mM) in ·NO scavenging ability, compared with a  $NO$  scavenging standard carboxy-PTIO (IC<sub>50</sub>:  $5.38 \pm 0.85$  mM).

#### Reaction of Hesperetin to ONOO<sup>-</sup>

To explore the scavenging mechanisms of hesperetin by which it undergoes either nitration reaction or electron donation after the addition of  $ONOO^{-}$ , spectrophotometric analysis at 400–450 nm was employed. Figure 2A,B show that tyrosine undergoes nitration by  $ONOO^-$ , because there is a peak around 430 nm. Hesperetin alone did not produce the peak around 430 nm (Fig. 2C), and in contrast to tyrosine, after interaction with  $ONOO^-$ , hespereting did not display any spectrometric change around 430 nm (Fig. 2D), strongly indicating that electron donation, but not nitration.

#### Effect of Hesperetin on BSA Nitration

Through Western blot analysis, we tested the ability of hesperetin to inhibit  $ONOO^-$ -mediated tyrosine nitration in common biological materials, such as BSA, using anti-3-nitrotyrosine antibody. The results indicated that hesperetin at concentrations of 2.5, 5, and  $10 \mu$ M attenuated the nitration of BSA in a dosedependent fashion, as shown in Fig. 3.

# Effects of Hesperetin on the Inhibition of Intracellular ROS Generation

As shown in Fig. 4A, the results revealed that preincubation of hesperetin at concentrations of 5, 15, and  $50 \mu M$  decreased the intracellular ROS generation induced *t*-BHP (20 $\mu$ M) in a dosedependent manner (Fig. 4A). As shown in Fig. 4B, compare of untreated control (Fig.  $4B(a)$ ),  $t$ -BHP  $(20 \mu M)$  and treated cells showed the generation ROS (Fig. 4B(b)). Pretreatment with hesperetin at concentrations of 5, 15, 50, and 200  $\mu$ M showed suppressed dose-dependent ROS generation (Fig. 4B(c–f)).



FIGURE 2 Interaction of hesperetin with ONOO<sup>-</sup>. Tyrosine (200  $\mu$ M) was incubated without (A) or with ONOO<sup>-</sup> (200  $\mu$ M) (B) in 50 mM phosphate buffer at pH 7.0 at room temperature for 10 min. Hesperetin (100  $\mu$ M) was incubated without (C) or with (D) ONOO<sup>-</sup> (200  $\mu$ M) at 378C for 10 min. The spectrophotometric analysis was performed as described in "Materials and methods" section. The arrow indicates a nitration peak at around 400–450 nm.

# Ability of Hesperetin to Inhibit Intracellular ONOO<sup>-</sup> Generation

As shown in Fig. 5A, the results revealed that preincubation of hesperetin at concentrations with 5, 15, and  $50 \mu$ M that intracellular ONOO<sup>-</sup> generation induced by  $t$ -BHP (20 $\mu$ M) decreased in a dosedependent manner (Fig. 5A). As shown in Fig. 5B,  $t$ -BHP (20 $\mu$ M) treated cells generated significantly higher  $ONOO^-$  levels than the untreated group (Fig. 5B(a,b)). Pretreated with hesperetin at concentrations of 5, 15, 50, and 200  $\mu$ M decreased the generation of  $ONOO^-$  in a dose-dependent manner (Fig. 5B(c–f)).



FIGURE 3 Effect of hesperetin on the nitration of BSA by ONOO<sup>-</sup>. Hesperetin and ONOO<sup>-</sup> were added to BSA. The reaction samples were incubated with shaking at 37°C for 20 min, which was resolved in 8% polyacrylamide-gel electrophoresis.

#### DISCUSSION

ONOO<sup>-</sup> is an important contributor to inflammatory conditions, including arthritis, endotoxic or septic shock, which is widely considered a systemic inflammatory disease, as well as being suspect in a variety of other models of inflammation.<sup>[26]</sup> Moreover, ONOO<sup>-</sup> can nitrate tyrosine, a process specific to this species and not triggered by the hydroxyl radicals, NO or  $O_2^{-}$   $[27]$  These are a variety of additive or synergistic cytotoxic processes triggered by  $ONOO^{-}$ , the combination of which can lead to acute and delayed cytotoxicity.<sup>[28]</sup> Plant food-derived antioxidants and active principles, such as flavonoids, hydrocinnamates (chlorogenic acid, ferulic acid, vanillin, etc.),  $\beta$ -carotene, and other carotenoids, vitamin E, vitamin C, rosemary, sage, and tea are important dietary antioxidant substances<sup>[29]</sup> that may provide efficient  $ONOO^-$  scavenger activities. Hesperetin is a widely known component of the plant kingdom that has been reported to have an antioxidant property.[30]

Our study shows that hesperetin is not only a strong ONOO<sup>-</sup> scavenger, but also effectively



FIGURE 4 Hesperetin inhibited intracellular ROS generation induced t-BHP. Cells were pre-treated with various concentrations of hesperetin for 2 h and further treated with *t-BHP (20 µM)* for 1 h. (A) Detection of intracellular ROS by staining with fluorometer using<br>DCFDA. Each value is the mean ± SEM. Statistical results of one-factor ANOVA: \*\*p vs. treated t-BHP. (B) The levels of intracellular ROS were detected by DCFDA using fluorescence microscope. (a) untreated control, (b) 20  $\mu$ M *t*-BHP treated alone, (c) 20  $\mu$ M *t*-BHP and 5  $\mu$ M hesperetin, (d) 20  $\mu$ M *t*-BHP and 15  $\mu$ M hesperetin, (e) 20  $\mu$ M *t*-BHP and 50  $\mu$ M hesperetin, (f)  $20 \mu M$  t-BHP and  $200 \mu M$  hesperetin.

inhibits intracellular free radical activity. It would seem reasonable to use the well-known ONOO<sup>-</sup> NO, and  $\cdot$ O<sub>2</sub> scavenging agents penicillamine,<sup>[31]</sup> carboxy–PTIO,<sup>[32]</sup> and trolox<sup>[33]</sup> in a comparison study against the scavenging activity of hesperetin. The most important findings from this present study are that the active component, hesperetin, not only directly enabled ONOO<sup>-</sup> scavenging, but also was involved in the inhibition of  $\overline{O_2}$  and  $\overline{O_2}$  and  $\overline{O_2}$ 

formation. These phenomena might be explained in terms of chemical structures. Based on hesperetin's structural similarity with other known antioxidants, including hydroxyl groups, hesperetin's hydroxyl structure may give the clue to its  $ONOO^{-}$  scavenging ability.

The minute mechanisms of  $ONOO^-$  scavenging actions by active components are not yet fully defined; so far, two possible pathways, nitration or



FIGURE 5 Effect of hesperetin on t-BHP mediated ONOO<sup>-</sup> generation. Cells were pre- treated with various concentrations of hesperetin for 2h and then treated with t-BHP (20  $\mu$ M) for 1 h. (A) Detection of intracellular ONOO<sup>-</sup> by staining with fluoremeter using DHR 123. Each value is the mean  $\pm$  SEM. Statistical results of one-factor ANOVA: \*\*p  $<$  0.01 vs. not treated t-BHP,  $^{\#}p$   $<$  0.05,  $^{\#}p$   $<$  0.01,  $^{\# \#}p$   $<$  0.001 vs. treated t-BHP. (B) The levels of intracellular ONOO<sup>-</sup> were detected by DHR 123 using fluorescence microscope. (a) untreated control, (b) 20  $\mu$ M t-BHP treated alone, (c) 20  $\mu$ M t-BHP and 5  $\mu$ M hesperetin, (d) 20  $\mu$ M t-BHP and 15  $\mu$ M hesperetin, (e) 20  $\mu$ M t-BHP and 50  $\mu$ M hesperetin, (f)  $20 \mu M$  t-BHP and  $200 \mu M$  hesperetin.

electron donation, have been suggested as a phenolic  $ONOO^-$  scavenger interaction with  $ONOO^-$ .<sup>[34]</sup> The phenolic compound, especially with a monohydroxyl group such as the phenolic amino acid, tyrosine, is preferentially nitrated by  $ONOO^{-}$ . Whether nitration is derived from the breakdown of  $ONOO^-$  to nitrogen dioxide radical ( $NO<sub>2</sub>$ ) or to the nitronium ion  $\widetilde{(NO_2^+)}$  has not clarified.<sup>[31]</sup>

In the present study, we exposed hesperetin to  $ONOO^-$  to determine whether hesperetin undergoes the nitration reaction. The addition of  $ONOO$ revealed no spectral change in the visible region, which proposed that the nitration of the aromatic ring did not occur. Furthermore, incubation of tyrosine and hesperetin with  $ONOO^-$  caused a decreased peak at 430 nm, which gave further

evidence supporting the possibility of an electron donation reaction between hesperetin and  $ONOO^-$ . Although the intermediate form of hesperetin after electron donation is not fully identified, it seems that the intermediate form was not toxic, because hesperetin treatment up to  $200 \mu M$  did not change YPEN-1 cell viability (data not shown).

The inhibitory action of hesperetin was further tested using additional physiological substances such as BSA. We found that  $ONOO^-$  could induce a functional damage in some biological molecules, such as BSA and LDL, via nitrotyrosine.<sup>[24]</sup> Protein tyrosine nitration by  $ONOO^-$  may interfere with phosphrylation/dephsphorylation signaling pathways and alter enzyme functions.<sup>[35-37]</sup> Nitrotyrosine has been reported in various hypertensive disorders, neurologic disorders, and chronic renal disease as a footprint of  $ONOO^{-}$ . [38-40] In the current study, evidence from Western blot analysis showed that hesperetin, even at a concentration as low as  $2.5 \mu M$ , could markedly reduce the nitrotyrosine present in BSA. With increases in hesperetin concentrations, BSA nitration decreased, further suggesting the putative anti- $ONOO^{-}$  action in vivo.

It has been reported that t-BHP stimulates the redox status of the cell and has been extensively utilized to elucidate the nature of oxidative stress in cellular physiology and biochemistry.<sup>[41]</sup>  $t$ -BHP generates ROS that are responsible for lipid peroxidation, DNA adduct formation, the induction of apoptosis. $[42]$  Therefore, in our study, cells were treated with t-BHP to induce oxidative stress. When treated with hesperetin, we found it inhibited the generation of ROS and  $ONOO^-$  in a dosedependent manner.

In summary, the present study suggested that the active bioflavonoid component, hesperetin, can scavenge  $ONOO^-$  efficiently. Hesperetin treatment led to an inhibition of ONOO<sup>-</sup>-mediated nitration of tyrosine through electron donation, and it also showed significant dose-dependent inhibition of BSA nitration from ONOO. Hesperetin was shown to t-BHP-induced free radical equally well. In conclusion, the current results demonstrated that hesperetin is an efficient  $ONOO^-$  scavenger and therefore potentially useful in the prevention of the ONOO2-related diseases, such as Alzheimer's disease, rheumatoid arthritis, cancer, atherosclerosis, and other inflammatory conditions.

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#### References

- [1] Mouly, P.P.M., Arzouyan, C.G., Gaydou, E.M. and Estienne, J.M. (1994) "Differentiation of citrus juices by factorial discriminate analysis using liquid chromatography of flavanone glycosides", J. Agric. Food Chem. 42, 70–79.
- [2] Rouseff, R.L., Martin, S.F. and Youtsey, C.O. (1987) "Quantitative survey of narirutin, naringin, hesperidin and neohesperidin in citrus", J. Agric. Food Chem. 35, 1027–1030.
- [3] Middeton, E. and Kanadaswami, C. (1992) "Effects of flavonoids on immune and inflammatory cell functions", Biochem. Pharmacol. 43, 1167–1179.
- [4] Kimura, Y., Kubo, M., Tani, T., Arichi, S. and Okuda, H. (1981) "Studies on scutellariae radix. IV. Effects on lipid peroxidation in rat liver", Chem. Pharm. Bull. 29, 2610–2617.
- [5] Bracke, M.E., Bruyneel, E.A., Vermeulen, S.J., Vennekens, K.I., Marck, V.V. and Mareel, M.M. (1994) "Citrus flavonoid effect on tumor invasion and metastasis", Food Technol. 48, 121–124.
- [6] Middeton, E. and Kanadaswami, C. (1994) "Potential healthpromoting properties of citrus flavonoids", Food Technol. 48, 115–119.
- [7] Attaway, J.A. (1994) "Citrus juice flavonoids with anticarcinogenic and antitumor properties", In: Huang, M.T., Osawa, T., Ho, C.T. and Rosen, R.T., eds, Food Phytochemicals for Cancer Prevention. I. Fruits and Vegetables (American Chemical Society, Washington, DC), pp 240–248.
- [8] Budavari, S., O'Neil, M.J. and Smith, A. (1985) "Merch Index", 11th ed. (Merck and Co., New Jersey), p 738.
- [9] Kim, A.R., Zou, Y., Kim, H.S., Choi, J.S., Chang, G.Y., Kim, Y.J. and Chung, H.Y. (2002) "Selective peroxynitrite scavenging activity of 3-methyl-1,2-cyclopentanedione from coffee extract", J. Pharm. Pharmacol. 54, 1385–1392.
- [10] Klotz, L.O., Schroeder, P. and Sies, H. (2002) "Peroxynitrite signaling: receptor tyrosine kinases and activation of stress-responsive pathways", Free Radic. Biol. Med. 33, 737–743.
- [11] Marletta, M.A., Hurshman, A.R. and Rusche, K.M. (1998) "Catalysis by nitric oxide synthase", Curr. Opin. Chem. Biol. 2, 656–663.
- [12] Beckman, J.S., Beckman, T.W., Chen, J., Marshall, P.A. and Freeman, B.A. (1990) "Apparent hydroxyl radical production by peroxynitrite: implication for endothelial injury from nitric oxide and superoxide", Proc. Natl Acad. Sci. USA 87, 1620–1624.
- [13] Pryor, W. and Squadrito, G. (1996) "The chemistry of peroxynitrite: a product from the reaction of nitric oxide with superoxide", Am. J. Physiol. 268, L699–L722.
- [14] Szabo, C. (1996) "The role of peroxynitrite in the pathophysiology of shock, inflammation and ischemia-reperfusion injury", Shock 6, 79–88.
- [15] Klotz, L.O. and Sies, H. (2003) "Defenses against peroxynitrite: selenocompounds and flavonoids", Toxicol. Lett. 140– 141, 125–132.
- [16] Virag, L., Szabo, E., Gergely, P. and Szabo, C. (2003) "Peroxynitrite-induced cytotoxicity: mechanism and opportunities for intervention", Toxicol. Lett. 140–141, 113–124.
- [17] Greenacre, S.A.B. and Ischiropoulos, H. (2001) "Tyrosine nitration: localization, quantification, consequences for protein function and signal transduction", Free Radic. Res. 34, 541–581.
- [18] Kong, S.K., Yim, M.B. and Stadtman, E.R. (1996) "Peroxynitrite disables the tyrosine phosphorylation regulatory mechanism: lymphocyte-specific tyrosine kinasa fails to phosphorylate nitrated cdc2(6–20) NH2 peptide", Proc. Natl Acad. Sci. USA 93, 3377–3382.
- [19] Grune, T., Klotz, L.O., Gieche, J., Rudeck, M. and Sies, H. (2001) "Protein oxidation and proteolysis by the nonradical oxidants singlet oxygen or peroxynitrite", Free Radic. Biol. Med. 30, 1243–1253.
- [20] Hajjar, D.P. and Harberland, M.E. (1997) "Lipoprotein trafficking in vascular cells: molecular trojan horses and cellular saboteurs", J. Biol. Chem. 272, 22975–22978.
- [21] Fici, G.J., Althaus, J.S. and VonVoigtlander, P.F. (1997) "Effects of lazaroids and a peroxynitrite scavenger in

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a cell model of peroxynitrite toxicity", Free Radic. Biol. Med. 84, 1127–1129.

- [22] Sies, H. and Masumoto, H. (1997) "Ebselen as a glutathione peroxidase mimic and as a scavenger of peroxynitrite", Adv. Pharmacol. 38, 2229–2246.
- [23] Kooy, N.W., Royall, J.A., Ischiropulos, H. and Beckman, J.S. (1994) "Peroxynitrite-mediated oxidation of dihydrorhodamine 123", Free Radic. Biol. Med. 16, 149–156.
- [24] Chung, H.Y., Choi, H.R., Park, H.J., Choi, J.S. and Choi, W.C. (2001) "Peroxynitrite scavenging and cytoprotective activity of 2,3,6-tribtomo-4,5-dihydroxybenzyl methyl ether from the marine alga Symphyocladia latiiuscula", J. Agric. Food Chem. 49, 3614–3621.
- [25] Nagata, N., Momose, K. and Ishida, Y. (1999) "Inhibitory effects of catecholamines and anti-oxidants on the fluorescence reaction of 4, 5-diaminofluorescein, DAF-2, a novel indicator of nitric oxide", J. Biochem. (Tokyo) 125, 658–661.
- [26] Szabo, C. (2003) "Multiple pathways of peroxynitrite cytotoxicity", Toxicol. Lett. 140–141, 105–112.
- [27] Crow, J.P. and Beckman, J.S. (1995) "Reaction between nitric oxide, superoxide, and peroxynitrite: footprints of peroxynitrite in vivo", Adv. Pharmacol. 34, 17–43.
- [28] Cuzzocrea, S., Zingarelli, B., O'Connor, M., Salzman, A.L. and Szabo, C. (1998) "Effect of L-buthionine-(S,R)-sulphoximine, an inhibitor of gamma-glutamylcysteine synthetase on peroxynitrite- and endotoxic shock-induced vascular failure", Br. J. Pharmacol. 123, 525–537.
- [29] Sies, H., Klotz, L.O., Sharov, V.S., Assmann, A. and Briviba, K. (1998) "Protection against peroxynitrite by selenoproteins", Z. Naturforsch C 53, 228–232.
- [30] Jung, H.A., Jung, M.J., Kim, J.Y., Chung, H.Y. and Choi, J.S. (2003) "Inhibitory activity of flavonoids from Prunus davidiana and other flavonoids on total ROS and hydroxyl radical generation", Arch. Pharm. Res. 26, 809-815.
- [31] Pannala, A.S., Rice-Evans, C., Halliwell, B. and Singh, S. (1997) "Inhibition of peroxynitrite-mediated tyrosine nitration by catechin polyphenols", Biochem. Biophys. Res. Commun. 232, 164–168.
- [32] Yoshida, M., Akaike, T., Goto, S., Takahashi, W., Inadome, A., Yono, M., Seshita, H., Maeda, H. and Ueda, S. (1998) "Effect of the NO scavenging carboxy-ptio on endothelium-dependent vasorelaxation of various blood vessels from rabbits", Life Sci. 62, 203–211.
- [33] Satoh, K., Kadofuku, T. and Sakagami, H. (1997) "Effect of trolox, a synthetic analog of  $\alpha$  -tocopherol, on cytotoxicity induced by UV irradiation and antioxidants", Anticancer Res. 17, 2459–2463.
- [34] Pannala, A., Razaq, R., Halliwell, B., Singh, S. and Rice-Evans, C.A. (1998) "Inhibition of peroxynitrite dependent nitration by hydroxycinnamates: nitration or electron donation?", Free Radic. Biol. Med. 24, 594–606.
- [35] Li, X., De Sarno, P., Song, L., Beckman, J.S. and Jope, R.S. (1998) "Peroxynitrite modulates tyrosine phosphorylation and phosphoinositide signaling in human neuroblastoma SH-SY5Y cells: attenuated effects in human 1321N1 astrocytoma cells", Biochem. J. 331, 599–606.
- [36] MacMillan-Crow, L.A. and Thompson, J.A. (1999) "Tyrosine modifications and inactivation of active site manganese superoxide dismutase mutant (Y34F) by peroxynitrite", Arch. Biochem. Biophys. 366, 82–88.
- [37] Vider, R.I., Ferrington, D.A., Williams, T.D., Bigelow, D.J. and Schoneich, C. (1999) "Protein modification during biological aging: selective tyrosine nitration of the SERCA2a isoform of the sarcoplasmic reticulum Ca2+-ATPase in skeletal muscle", Biochem. J. 340, 657–669.
- [38] Vaziri, N.D., Liang, K. and Ding, Y. (1999) "Increased nitric oxide inactivation by reactive oxygen species in lead-induced hypertension", Kidney Int. 56, 1492-1498.
- [39] Roberts, C.K., Vaziri, N.D., Wang, X.Q. and Barnard, R.J. (2000) "Enhanced NO inactivation and hypertension induced by a high-fat, refined-carbohydrate diet", Hypertension 36, 423–429.
- [40] Deng, G., Vaziri, N.D., Jabbari, B., Ni, Z. and Yan, X.X. (2001) "Increased tyrosine nitration of the brain in chronic renal insufficiency: reversal by antioxidant therapy and angiotensin-converting enzyme inhibition", J. Am. Soc. Nephrol. 12, 1892–1899.
- [41] Elliott, S.J., Eskin, S.G. and Schilling, W.P. (1989) "Effect of t-butyl-hydroperoxide on bradykinin-stimulated changes in cytosolic calcium in vascular endothelial cells", J. Biol. Chem. 264, 3806–3810.
- [42] Haidara, K., Morel, I., Abalea, V., Gascon, B.M. and Denizeau, F. (2002) "Mechanism of tert-butylhydroperoxide induced apoptosis in rat hepatocytes: involvement of mitochondria and endoplasmic reticulum", Biochim. Biophys. Acta 1542, 173–185.